



The Impact of Geochemistry

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As geochemists and mineralogists, we are well aware of the impact of our science. We can often reel off examples of how our discoveries have influenced industry and improved humankind's knowledge about how the Earth works, as well as how natural and anthropogenic processes have led to its present state. Moreover, because of a century of analytical developments and conceptual breakthroughs, geochemists are now versatile and can now work across the entire spectrum of the Earth sciences. However, we are not so good at promoting the social and economic impacts of geochemistry.

KEYWORDS: economic, social, impact, communication, geochemistry

EXAMPLES OF GEOCHEMICAL IMPACT

This issue of *Elements* demonstrates the application of geochemistry to a variety of societally and economically important areas, including, (1) exploration for and sustainable exploitation of mineral deposits; (2) environmental mineralogy, the interaction of microbial organisms with minerals and the close links between minerals and materials science, particularly in the areas of energy-related materials and nuclear waste disposal; (3) the use of geochemistry as a guide to solving environmental problems in cities, using London (England) as a case study; (4) the use of stable isotopes to detect adulterated foods and beverages and; (5) to detect and monitor the progress of diseases in humans. A significant driver for the research described in all of these articles is analytical achievement and translating this to a societal application. There is no doubt that academic and government geochemists working together with the analytical industry have pushed the limits of discovery-mode science, which has led to applications that are routinely used to benefit society and the economy.

The papers in this issue focus on the prosperity of our society and our standard of living. Our quality of life is directly related to our ability to find, exploit, and manage our metal and mineral resources. The first in this series of papers (Kyser et al. 2015) points out that metal and mineral deposits are, in fact, geochemical anomalies and, as such, applied geochemistry plays a critical role throughout the mineral resources value chain from early-stage exploration to mine closure. Minerals also help us design new materials for advanced technologies, ranging from energy production to management of contaminated areas and the ultimate

disposition of wastes. The second paper (Calas et al. 2015) discusses evidence of the environmental importance of nanomaterials, non- and poorly crystalline phases and interactions between minerals and ubiquitous microbial activity. It is probable that in both of these cases geochemistry will lead to further economic advantages to ensure a more sustainable planet with cleaner exploration strategies and mineral extraction and to better environmental cleanup activities.

The challenges and importance of communication between scientists and policy makers are featured in the third paper (Ludden et al. 2015). This paper also examines how the application of geochemical tools and approaches can provide evidence to underpin decision-making, as well as defining solutions to environmental problems in cities; in this case London and its surroundings. The paper questions our ability to move from the provision of geochemical evidence to describing a solution and then convincing politicians to put this into practice.

Ehrlinger et al. (2015), in the fourth paper, demonstrate how the classic geochemical technique of isotopic tracing and technology can be applied to two sectors of relevance to society: food authenticity and public safety. One involves the use of stable isotopes as biogeochemical markers, which can allow us to tell if food and beverages have been adulterated and so protect consumers from fraud. A second theme involves stable isotopes in areas of interest to law enforcement to determine origins and travel histories of unidentified murder victims and tracing the origins of illicit drugs. Such advances in forensics could only have been achieved by expansion from the traditional applications of mass spectrometry.

The final paper in the series (Albarède 2015) discusses a highly innovative and promising area for geochemistry: diagnosing and monitoring the progression of disease in humans. The conventional stable isotope systems of C, H, O, N, and S are rarely used for medical purposes because they lack specificity. However, new medical applications of less conventional stable isotopes are arising, such as Fe in heme cores, Cu in erythropoiesis (red blood cell production), and Ca in bones. These “unconventional” isotopes are increasingly being used because they have a much smaller number of specific functional roles in biology and because they have a shorter turnover rate in the body. The whole periodic table is now opening up to medical and environmental applications thanks to isotope geochemistry.

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